organic papers

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Key indicators

Single-crystal X-ray study T = 90 K Mean σ (C–C) = 0.003 Å R factor = 0.034 wR factor = 0.077 Data-to-parameter ratio = 10.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(*Z*)-2-(4-Methoxybenzylidene)-1-azabicyclo[2.2.2]octan-3-one

The title compound, $C_{15}H_{17}NO_2$, was prepared by the basecatalyzed reaction of 4-methoxybenzaldehyde with 1-azabicyclo[2.2.2]octan-3-one. The configuration about the olefinic bond connecting the methoxyphenyl and 1-azabicylo[2.2.2]octan-3-one moieties is Z. Received 15 December 2005 Accepted 20 December 2005 Online 7 January 2006

Comment

The title compound, (I), was prepared by the base-catalyzed condensation of 4-methoxybenzaldehyde with 1-azabicyclo[2.2.2]octan-3-one, to afford (I) as a single geometrical isomer. In order to confirm the double-bond geometry, and to determine how the molecular conformation in the crystal structure is affected by the position of the *para*-methoxy group, the X-ray analysis of this positional isomer has been carried out and the results are presented here. This is a companion study together with the previous communication on the isomeric 2-methoxy analogue (Sonar *et al.*, 2006).

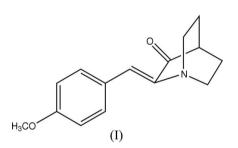


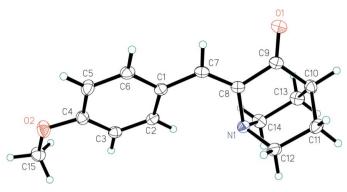
Fig. 1 illustrates an ellipsoid plot of (I), with the atomnumbering scheme; selected geometric parameters are listed in Table 1. The configuration about the olefinic bond connecting the 4-methoxyphenyl and 1-azabicylo[2.2.2]octan-3-one moieties is Z. The double bond has a nearly planar atomic arrangement, since the r.m.s. deviation from the mean plane passing through atoms N1, C8, C9, C7 and C1 for (I) is 0.0197 (11) Å.

There are no significant differences in the geometric parameters of (Z)-2-(2-methoxy-benzylidene)-1-azabicyclo[2.2.2]octan-3-one and (Z)-2-(4-methoxy-benzylidene)-1azabicyclo[2.2.2]octan-3-one. This suggests that the position of the methoxy group does not have much influence on the overall molecular conformation in the 2- and 4-positional isomers.

Experimental

Compound (I) was prepared following the method described previously for the 2-methoxy analogue (Sonar *et al.*, 2006), but utilizing 4-methoxybenzaldehyde in place of 2-methoxy-

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A view of the molecule of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small spheres of arbitrary radii.

benzaldehyde. Spectroscopic analysis: ¹H NMR (CDCl₃, δ , p.p.m.): 1.99–2.04 (*td*, 4H), 2.59–2.62 (*p*, 1H), 2.93–3.03 (*m*, 2H), 3.09–3.19 (*m*, 2H), 3.83 (*s*, 3H), 6.89 (*dd*, 2H),6.98 (*s*, 1H), 8.02 (*dd*, 2H); ¹³C NMR (CDCl₃, δ , p.p.m.): 26.4, 40.6, 47.8, 55.5, 114.1, 125.1, 127.0, 134.1, 143.0, 160.8, 206.4.

Crystal data

C₁₅H₁₇NO₂ $M_r = 243.30$ Orthorhombic, $P2_12_12_1$ a = 5.8425 (2) Å b = 9.9252 (3) Å c = 21.3739 (7) Å V = 1239.43 (7) Å³ Z = 4 $D_x = 1.304$ Mg m⁻³

Data collection

Nonius KappaCCD area-detector diffractometer ω scans Absorption correction: multi-scan (SCALEPACK; Otwinowski & Minor, 1997) T_{min} = 0.975, T_{max} = 0.987 10079 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.034$ $wR(F^2) = 0.077$ S = 1.041664 reflections 165 parameters H-atom parameters constrained Mo $K\alpha$ radiation Cell parameters from 1641 reflections $\theta = 1.0-27.5^{\circ}$ $\mu = 0.09 \text{ mm}^{-1}$ T = 90.0 (2) K Block, colourless $0.30 \times 0.20 \times 0.15 \text{ mm}$

1664 independent reflections 1323 reflections with $I > 2\sigma(I)$ $R_{int} = 0.031$ $\theta_{max} = 27.5^{\circ}$ $h = -7 \rightarrow 7$ $k = -12 \rightarrow 12$ $l = -27 \rightarrow 27$

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0341P)^{2} + 0.1346P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.20 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.18 \text{ e} \text{ Å}^{-3}$ Extinction correction: *SHELXL97* (Sheldrick, 1997) Extinction coefficient: 0.013 (2)

| Table 1 | _ | |
|----------------------------------|----|-----|
| Selected geometric parameters (A | Å, | °). |

| C1-C7 | 1.463 (2) | O2-C15 | 1.429 (2) |
|--------------|-------------|-------------|-------------|
| N1-C8 | 1.447 (2) | C7-C8 | 1.336 (2) |
| O1-C9 | 1.227 (2) | C8-C9 | 1.485 (2) |
| O2-C4 | 1.369 (2) | C9-C10 | 1.508 (3) |
| C2-C1-C7 | 123.56 (17) | C7-C8-C9 | 121.39 (17) |
| C6-C1-C7 | 118.35 (17) | N1-C8-C9 | 113.57 (15) |
| C4-O2-C15 | 117.91 (16) | 01-C9-C8 | 124.48 (17) |
| C8-C7-C1 | 130.35 (17) | C8-C9-C10 | 110.75 (15) |
| C15-O2-C4-C3 | -5.4 (3) | C6-C1-C7-C8 | 160.91 (19) |
| C2-C1-C7-C8 | -21.9(3) | C7-C8-C9-O1 | 0.0 (3) |

In the absence of significant anomalous dispersion effects, Friedel pairs were averaged. H atoms were positioned geometrically and treated as riding, with C–H distances in the range 0.95–0.99 Å and with $U_{\rm iso}({\rm H}) = 1.2-1.5U_{\rm eq}({\rm C})$.

Data collection: *COLLECT* (Nonius, 1999); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL/PC* (Sheldrick, 1995); software used to prepare material for publication: *SHELX97-2* (Sheldrick, 1997) and local procedures.

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